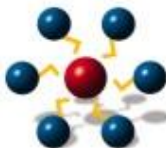
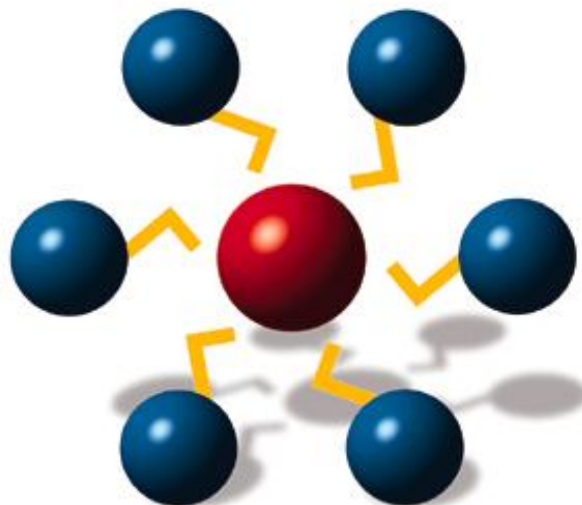
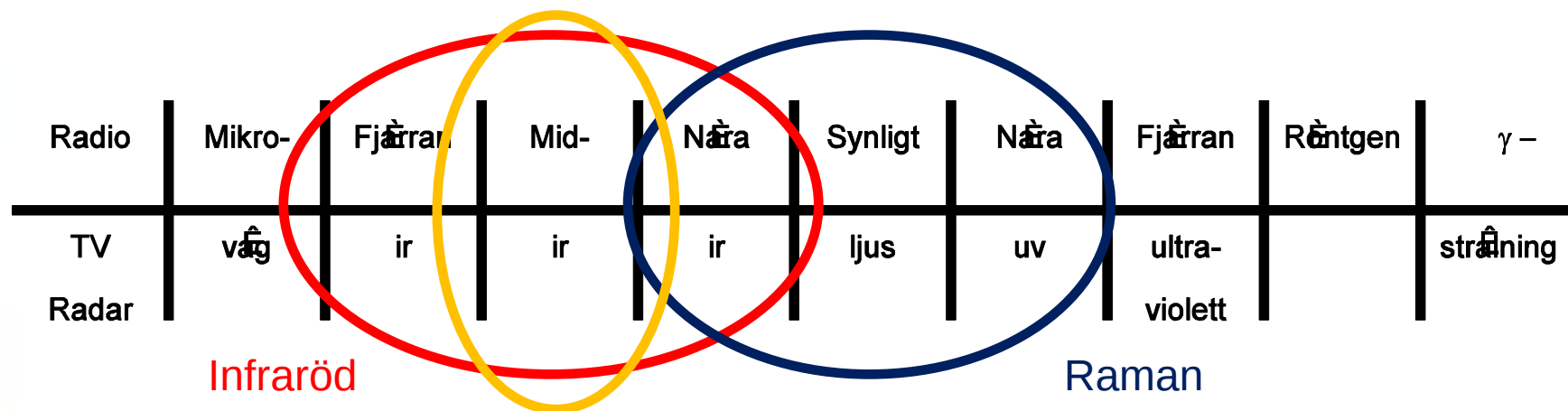


Vibrationspektrometri

Matti Hotokka



Elektromagnetiskt spektrum



λ	3 cm	1 mm	25 μm	2.5 μm	770 nm	400 nm	200 nm	10 nm	0.2 nm
$\tilde{\nu}$ (cm^{-1})	0.3	10	400	4000	13,000	25,000	50,000		
E(eV)	4×10^{-5}	10^{-3}	5×10^{-2}	0.5	2	3	6	100	6×10^3
E(J)	6×10^{-24}	2×10^{-22}	8×10^{-21}	8×10^{-20}	3×10^{-19}	5×10^{-19}	10^{-18}	2×10^{-17}	10^{-15}
E(Hz)	9×10^9	3×10^{11}	10^{13}	10^{14}	4×10^{14}	7×10^{14}	10^{16}	3×10^{16}	10^{18}
E(K)	0.4	14	6×10^2	6×10^3	2×10^4	4×10^4	7×10^4	10^6	7×10^7

NMR

Rotations-

Vibrations-

Elektroniska

Fotoelektron-

ESCA

EPR

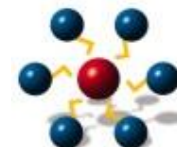
spektra

spektra

övergångar

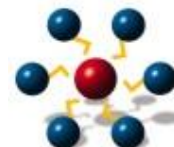
spektra

Auger



Energienheter

⇒ Vågtal (cm^{-1}); det vanliga	$\tilde{\nu}$
⇒ Frekvens (Hz, cps); för teoretiker	ν
⇒ Vinkelhastighet (1/s); föråldrad	ω
⇒ Våglängd (μm); sällsynt	λ
⇒ Joule (J); opraktisk	E



Infrarödområden

⇒ Fjärran ir (FIR, far infrared)

➤ $10 - 400 \text{ cm}^{-1}$

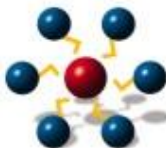
⇒ Mid-ir (MIR)

➤ $400 - 4000 \text{ cm}^{-1}$

⇒ Nära ir (NIR, near infrared)

➤ $4000 - 12\,000 \text{ cm}^{-1}$

⇒ Alla har användningar, Mid-ir viktigast



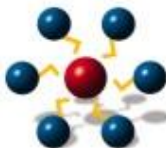
Mätmetoder

⇒ Infraröd

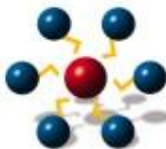
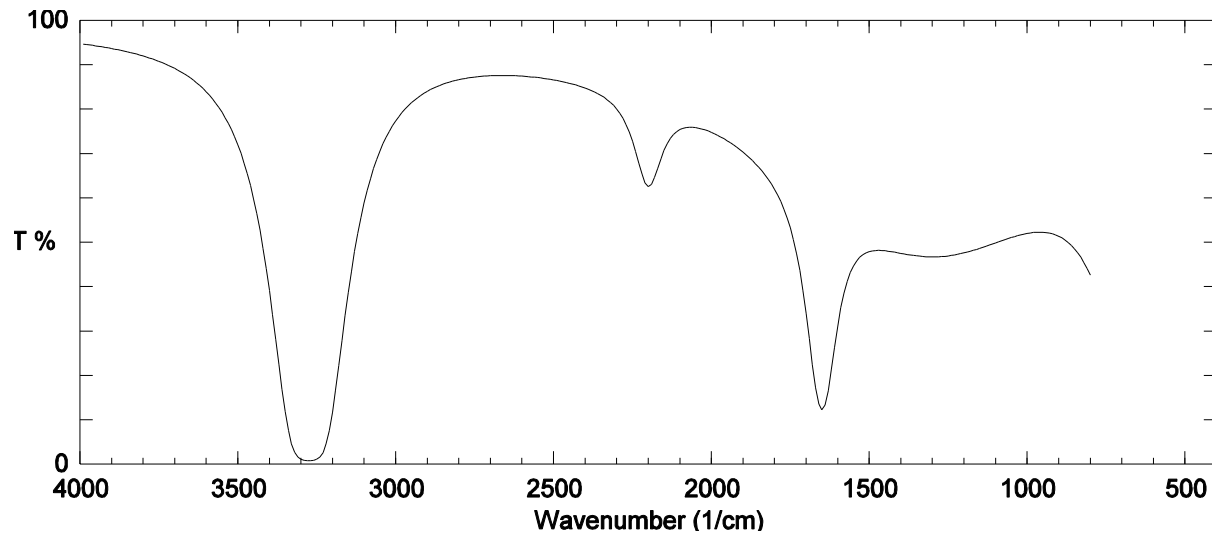
- Vanlig
- Baserar sig på absorption

⇒ Raman

- Sällsynt (tyvärr)
- Baserar sig på ljusspridning



Infrarödspektrum



Beer-Lambert

⇒ Absorptionshastighet

$$dI(\lambda) = -I(\lambda)\gamma(\lambda)dx$$

⇒ Intensitet efter provet

$$I(\lambda) = I_0(\lambda)e^{-\gamma(\lambda)\ell}$$

⇒ Transmittans

$$T(\lambda) = I(\lambda) / I_0(\lambda) = e^{-\gamma(\lambda)\ell} = e^{-\varepsilon(\lambda)c\ell}$$

⇒ Absorbans

$$A(\lambda) = \ln(I_0(\lambda) / I(\lambda)) = \varepsilon(\lambda)c\ell$$

